## **Baking Soda and Vinegar Explosion**

**Florida State Standard:** SC.5.P.8.2 – Recognize that gases are released when certain substances react with one another.

**Florida State Benchmark:** SC.5.P.8.3 – Investigate and describe how gas is produced in different reactions.

**Objective:** Students will observe and understand the chemical reaction between baking soda and vinegar, focusing on acid-base reactions and gas production. **Materials:** 

- Baking soda (sodium bicarbonate)
  - Vinegar (acetic acid)
  - Small container (cup or bowl)
  - Tray or larger container (to catch spills)
  - Optional: food coloring for visual effect
  - Safety goggles

## **Safety Precautions:**

Students must wear safety goggles to protect their eyes from potential splashes during the experiment. The experiment should be conducted in a well-ventilated area to avoid inhaling any strong odors from the vinegar. Ensure the workspace is clear of any valuable materials that could be damaged by spills. Handle all materials with care to avoid accidents.

## **Procedures:**

- 1. **Introduction (10 minutes):** Begin the lesson by discussing what a chemical reaction is and ask students if they have ever seen a chemical reaction before. Introduce the materials and explain that they will be observing the reaction between baking soda and vinegar.
- 2. **Experiment (20 minutes):** Have students put on safety goggles. Instruct them to place 1-2 tablespoons of baking soda into a small container. Slowly pour vinegar into the container and observe the reaction. If desired, students can add food coloring to the vinegar for a more visually exciting effect.
- 3. **Observation (10 minutes):** Ask students to describe the reaction. Encourage them to think about the formation of bubbles, changes in the mixture, and what gas might be produced.
- 4. **Generalization (10 minutes):** Explain the chemical reaction that took place: baking soda (a base) reacts with vinegar (an acid) to produce carbon dioxide gas, water, and sodium acetate. Discuss real-world examples of gas production, such as in baking or fermentation.
- 5. **Assessment (10 minutes):** Evaluate students' understanding through a brief quiz or a group discussion. Ask students to summarize the reaction in a short paragraph, explaining the roles of the reactants and the gas produced.

https://www.youtube.com/@STEMClub-z7l
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Note: Clean-up

Ensure students clean up any spills immediately using paper towels or rags. All materials should be disposed of properly, and containers should be rinsed out with water. Safety goggles should be returned to the appropriate storage area.